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[Ion Selective Theory - Ionode Electrodes](#)

An ion selective electrode is a sensor which converts the activity of a specific ion (dissolved in a solution) into a voltage (potential), which can be measured by a mV or Ion meter. The voltage is theoretically dependent on the logarithm of the ion activity, as described by the Nernst Equation.

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The term 'ion-selective electrodes' is applied to a range of membrane electrodes which respond selectively toward one (or several) ionic species in the presence of others.

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[Abstract. Potentiometric enzyme electrodes have been](#)

designed for the determination of several clinically relevant species (1-3). They typically consist of an enzyme layer covering the surface of an ion-selective electrode or a gas sensor which measures the products released by enzymatic degradation of substrates.